

Technical Report #23  
Trace Metals in the Gulf of Maine Determined by  
Differential Pulse Anodic Stripping Voltammetry  
by  
M.L. Brann and C.M. Yentsch

Bigelow Laboratory for Ocean Sciences  
McKown Point  
West Boothbay Harbor, Maine 04575

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**Key Words:** Trace metals; differential pulse anodic stripping voltammetry; untreated seawater; available trace metals; Gulf of Maine.

#### Abstract

The method of Differential Pulse Anodic Stripping Voltammetry is briefly explained along with its application to trace metal analysis of cadmium, lead, and copper in seawater. Data has been collected for the past few years at several locations in the Gulf of Maine. The results are displayed in a clear bar graph form. Because numerous measurements were taken at the same time, there is the potential of comparing and correlating trace metals with other parameters.

#### Introduction

Trace metals are an important factor in the marine environment. For many of these metals a small amount is necessary in the role as a nutrient, but larger amounts become toxic. Marine life reacts in different ways to this stress. While some forms can adjust in various ways, others cannot survive. The latter seems to be the case for *Gonyaulax tamerensis*, and thus trace metals are an important measurement in our study of the red tide.

The method of Differential Pulse Anodic Stripping Voltammetry (DPASV) was chosen for seawater trace metal analysis for many reasons. Probably the two most important reasons are: 1) this method can analyze only the "available" portion of the metals, i.e. the portion that marine life sees, and 2) the sample needs no pre-treatment, only to be frozen during storage. Very few chemicals are required and concentrations as low as tenths of ppb can be measured and reproduced.

The basic ideas behind DPASV are as follows: The trace metals positive of the deposition potential are reduced at the working

electrode, which is simply a mercury drop. The trace metal actually forms an amalgam with the mercury. At the end of the deposition time, the magnetic stirrer stops and the cell quiets down. Following this, the stripping phase begins. The instrument scans from the deposition (initial) potential to the final potential. As each trace metal's characteristic half potential is reached, it is oxidized back into solution, thereby releasing electrons which are processed by the instrument and recorded as peaks on the trace.

#### Materials and Methods

New nalgene LPE bottles are used for samples. The bottles are acid washed in 6N Nitric acid, then 6 N Hydrochloric acid, and finally clean, i.e. trace metal free, seawater. Each step takes at least 24 hours. After washing, the bottle is rinsed with copious amounts of ultrapure water.

The samples are collected from the ocean either by hand or from a plastic bucket, keeping as far away as possible from boat bottoms, chains on docks, etc. The bottle is rinsed five times before the final filling, capped and frozen immediately. The trace metal analysis is done with a polarographic analyzer. The instrumental set-up consists of: PAR model 174 Polarographic Analyzer; PAR model 315A Automated Electroanalysis Controller; PAR model 303 Static Mercury Drop Electrode; a Moseley model 7000 AMR X-Y Recorder; a voltage regulator; and a magnetic stirrer regulated by the line frequency.

Instrumental parameters vary with the samples, but here are some typical settings:

Model 174 - rate = 2 mv/s, scan direction = +, range = 1.5 volts,  
initial potential = -0.00 v, modulation amplitude = 100 mv, operating  
mode = diff. pulse, current range = 5 uA, display direction = -.  
Model 135 A - conditioning potential = - 0.00 V, deposition/initial  
potential = - 1.02 V, final potential = -0.08 V, equilibration = 15  
seconds, and the deposition time usually ranges between 100 and 500  
seconds with the maximum being 999 seconds.

One of the greatest advantages of trace metal analysis by ASV is  
the use of very few chemicals. These are: triple distilled mercury (or  
better), reagent grade nitric and hydrochloric acid, ultra pure water,  
ultra pure nitric acid, siliconizing fluid, carbon dioxide gas,  
saturated KCl, AgCl electrolyte for reference electrode, and 1000 ppm  
atomic absorption standards of cadmium, lead and copper.

A primary standard is made at least four times a year from 1000 ppm  
standards by diluting 2.5 ml of Cd and 25.0 ml each of Pb and Cu to 100  
ml in a volumetric flask. The secondary standard is made once a week  
from the primary standard by diluting 0.50 ml to 50 ml in a volumetric  
flask. Spiking is done with a 10 or 20  $\mu$ l micropipette.

The sample is thawed immediately prior to analysis. All three  
electrodes and purge tube are rinsed with 6 N nitric acid, then ultra  
pure water, dried with a Kimwipe, then about 50 drops of mercury are  
dispensed to help clear any air from the mercury column. A 10.0 ml  
sample is placed in a glass cell. The cells are soaked in 6 N nitric  
acid, then in clean un-acidified seawater, then in the sample itself  
before hand. Hopefully, by this point, the cell walls and sample have  
reached equilibrium, and trace metals will not be absorbed or released  
during the analysis. A teflon coated stirring bar which is soaked in

nitric acid when not in use is rinsed with the ultra pure water and placed in the cell. The cell is then loaded and purged for about ten minutes with  $\text{CO}_2$  to rid the sample of interfering oxygen. After purging is completed, deposition begins at a selected potential. This automatically continues for the selected amount of time, depending on the sample. This is followed by a 15 second equilibration and then a scan from -1.02 volts to -0.08 volts at a rate of 2 mV/s. During this stripping phase the trace metals are oxidized at their characteristic half potentials, slightly variable, appearing in order, Cd, Pb, and Cu. One mercury drop is used for each sample.

Concentrations are determined by the method of standard additions. Usually the spikes are 0.25 or 0.50 parts per billion (ppb) for Cd, and 2.5 or 5.00 ppb for Pb and Cu. Three standard additions are made. The final concentration is calculated using a linear regression program, plotting peak height against concentration of spikes, and shifting the calibration curve to pass through the origin. If the samples are from the same location and general time period, and if the shape of the traces is identical, then a standard curve is made in the morning and the rest of the samples for that day are calculated using that standard curve.

#### Results

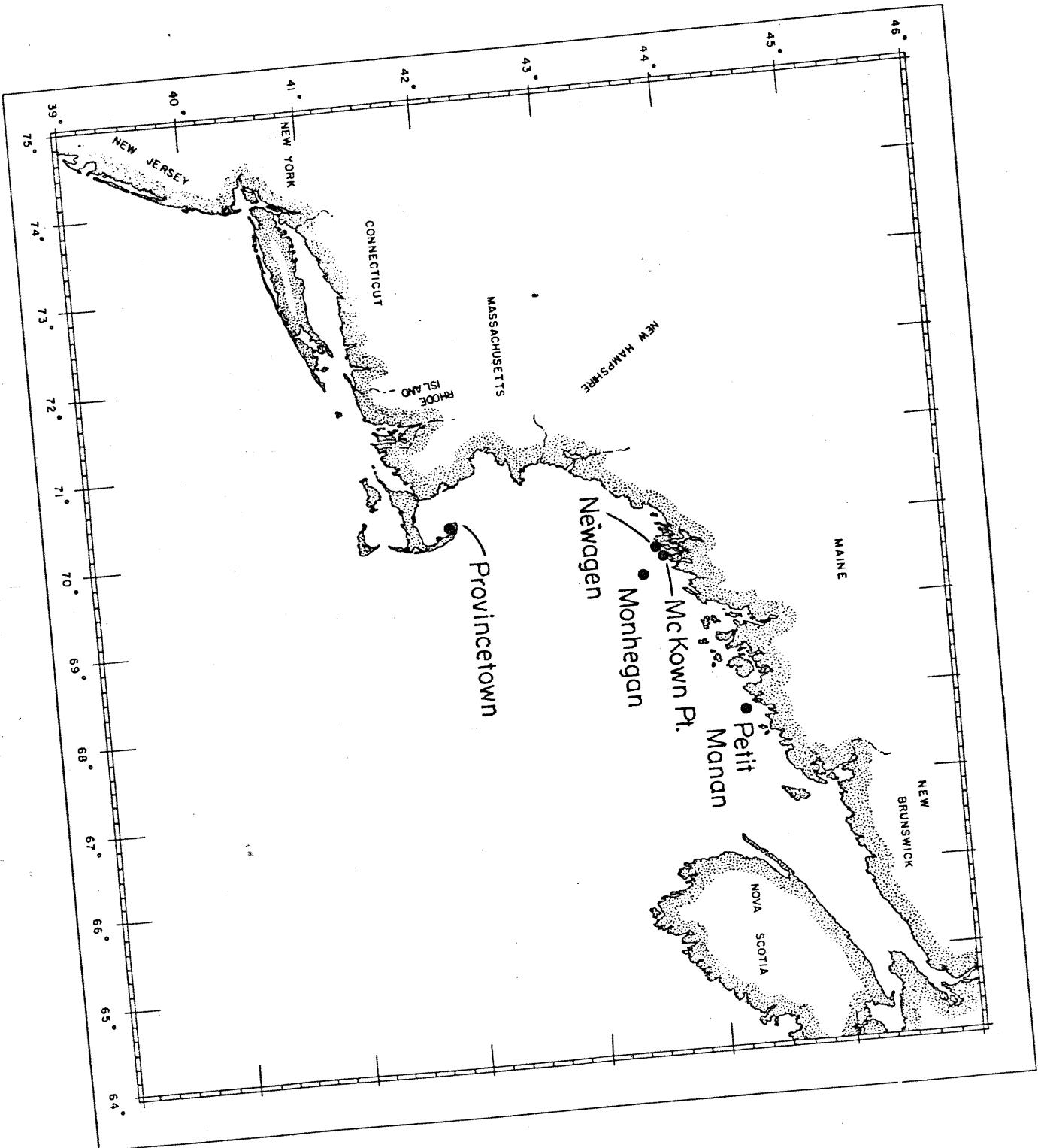
During the past three years, 1977-1979, several monitoring stations have been established. In 1977, Monhegan Island (about 15 miles SE of Boothbay Harbor, Maine) was the only station. In 1978, four more stations were added: Cape Newagen, near Boothbay Harbor; a station at the lab, in the harbor, called McKown Point; Petit Manan Point, east of Penobscot Bay; and Provincetown, Mass. In 1979, samples were taken at

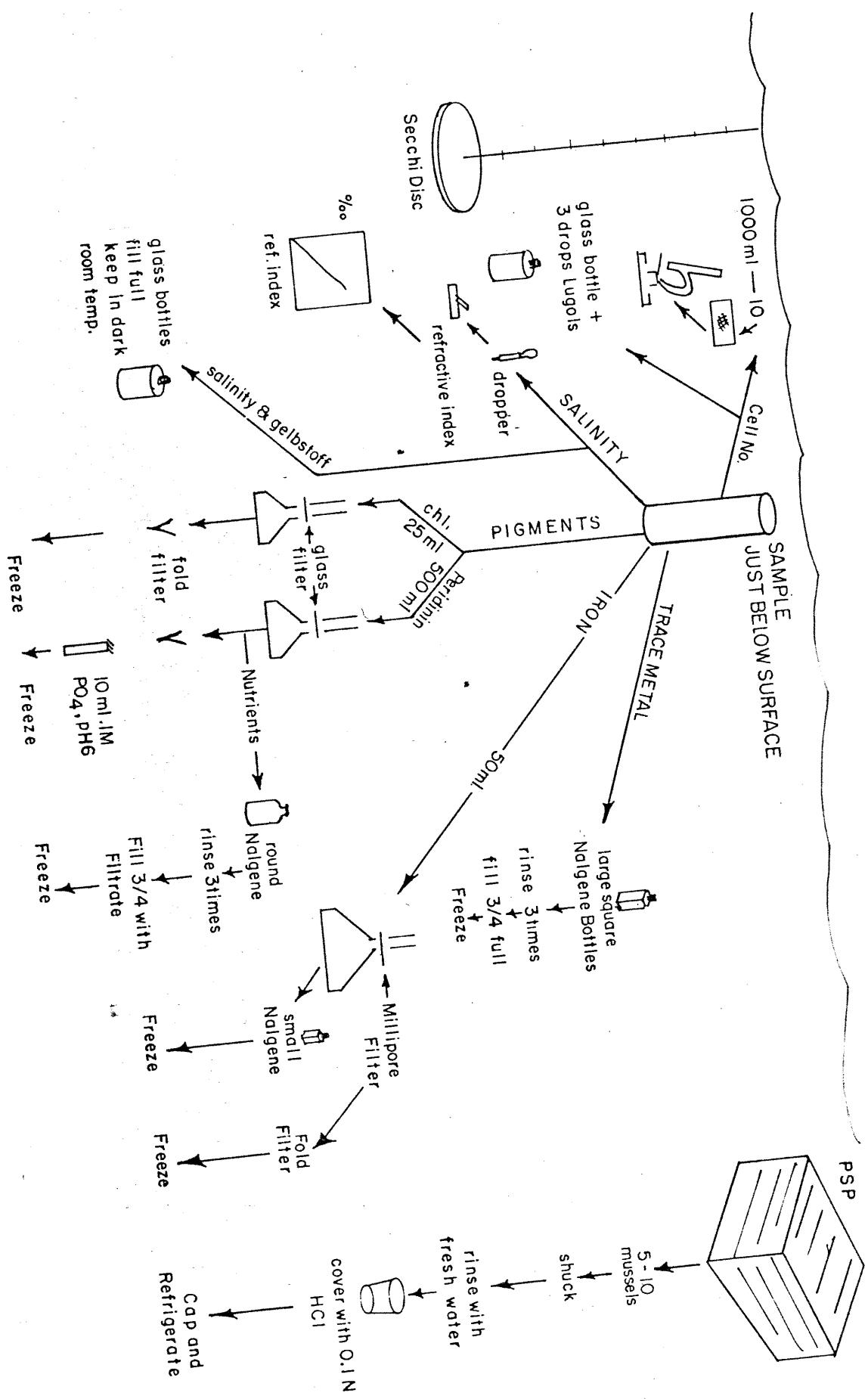
Monhegan Island, Cape Newagen, Petit Manan Point, and Provincetown. Provincetown and Newagen samples are now taken year round, the others only in the summer. Sampling locations are shown on the chart.

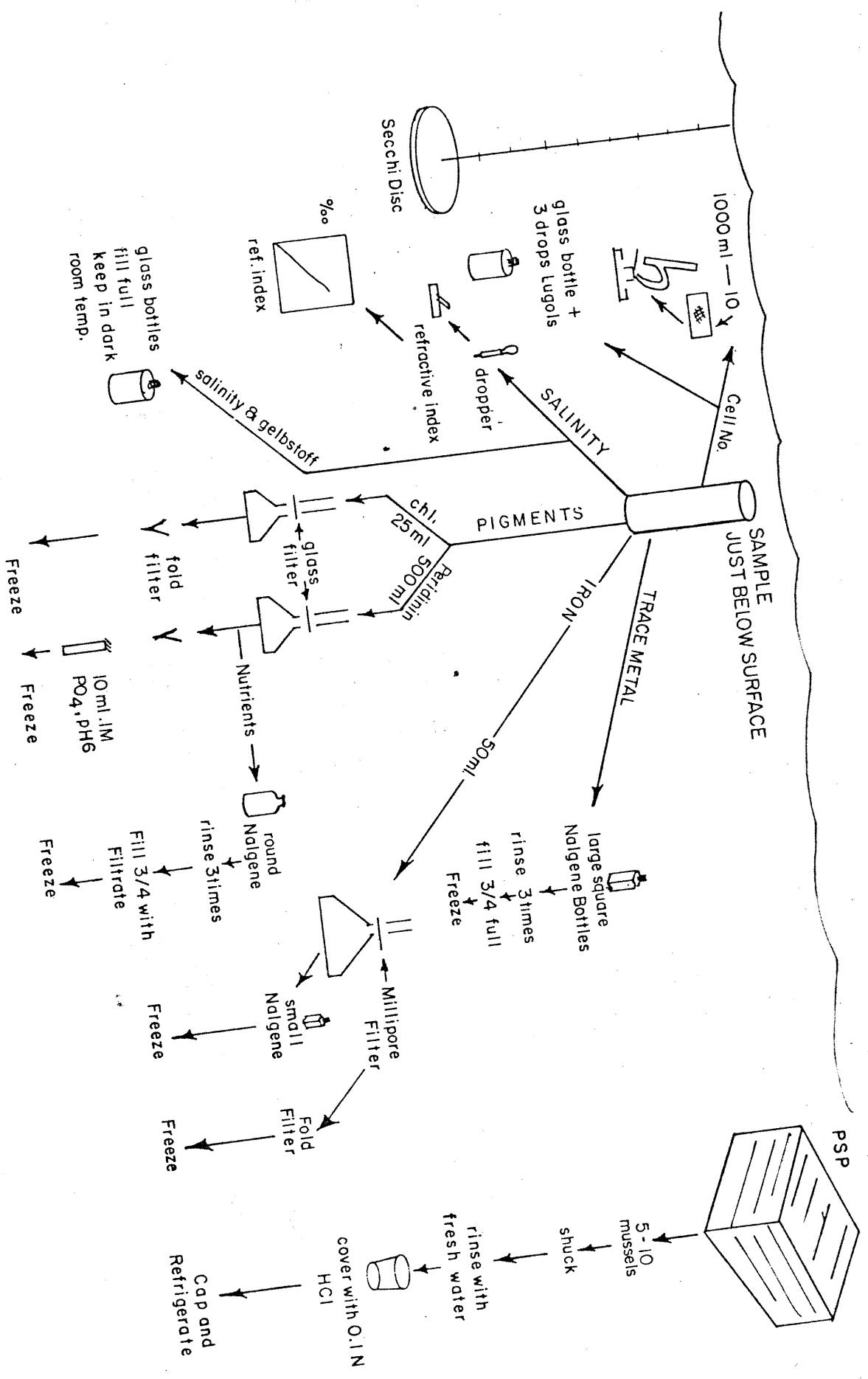
The samples are taken in the manner described previously. For a while samples were acidified, but it has been decided that the samples are to be frozen. Only frozen data is reported here. The data is also in the form of bar graphs for the three metals, cadmium, lead, and copper, for all stations for 1977, 1978 and part of 1979. All the samples are analyzed after purging with carbon dioxide, which gives a pH of about 4.8. Therefore, the concentrations reported are the metals that are available for reduction in untreated seawater at a pH of 4.8.

#### Discussion

This data is presented merely to show what sort of concentrations have been found in the Gulf of Maine with the method of DPASV. Concurrent with these samples, chlorophyll, salinity, temperature, nutrients, cell counts, some paralytic shellfish poison and iron measurements have been taken. The weather conditions and sea state are also available. Therefore, there is the opportunity to compare trace metal concentrations with other parameters. This has been done on a very limited basis with no significant results so far. It seems that trace metal ion concentrations would be influenced by such things as run off, upwelling, changes in organics, etc., with many factors simultaneously affecting the trace metals. In the future, organic measurements could also be made to compliment the trace metal data. Perhaps factors influencing trace metals can be established and the influence of trace metals on the marine environment defined.







Argument on why total concentration, not available, are obtained from the ASV

During the past 2½ years, the results obtained from the method of ASV have been reported as representing the available portion of the trace metals, not the total concentration. While it is true that only the available, or reducible trace metals are actually analyzed by this method, the final concentration calculated could be the total concentration. It seems unlikely that total concentrations could be obtained from an instrument that can only analyze available and this is why I never questioned it.

First, let me explain a few terms. The "available" portion is that fraction of the total trace metals that is either in an ionic form or loosely bound. Trace metals in these forms can be used by marine organisms, thus, the term available. For the same reason they are also reducible by the method of ASV. On the other hand, trace metals that are either tightly bound chemically to organics or physically trapped inside a colloidal particle cannot be used by marine organisms and also cannot be chemically reduced by ASV. Thus, they are called the non-reducible fraction. For the sake of clarity trace metals are now called either reducible or non-reducible.

Basically, concentrations are determined as follows. A scan is run on the sample (untreated seawater, purged with  $\text{CO}_2$ ,  $\text{pH} = 4.8$ ) and the peak heights represent the reducible fraction of trace metals in the sample itself. Say, for example, Cd has a peak height of 2. Then a known amount, a spike, say 0.25 ppb of  $\text{Cd}^{+2}$ , is added and a scan is run. This second peak height represents both the sample and the spike, but

again only the reducible fraction. Say Cd now has a peak height of 8. Subtracting the sample peak height from the spike gives the increase in peak height due to the spike. This increase which is  $8-2 = 6$ , has always been called 0.25 ppb because that is what was added. On the basis of this, the sample's concentration of Cd is calculated, which is  $0.25 \times 2/6 = 0.083$  ppb of Cd.

The problem with this stems from the fact that the peak increase of 6 does not really represent 0.25 ppb. When the spike is added some stays in the ionic form, and some form inorganic compounds, some of which are non-reducible. Thus, the peak height of 6 is due to the reducible fraction of the 0.25 ppb spike and really only represents say 0.20 ppb of Cd.

Now, assuming that equilibrium is reached rapidly and that the ratio of reducible to non-reducible trace metals is always the same, the following conclusions are reached. Say, for example, 20% of all Cd is non-reducible. That means that if the peak height were to represent total concentration then the sample peak height of Cd would now be  $2 + 20\%$  of 2 which is 2.4, and the spike would be  $8 + 20\%$  of 8 which is 9.6. Calculating the concentration as before:  $9.6-2.4 = 7.2$  (spike height alone) and  $0.25\text{ppb} \times \frac{2.4}{7.2} = 0.083$  ppb of Cd. The result is obviously total but yet it is the same as the first calculation. The reason for this is that the ratio of sample to spike is the same whether the peaks are reducible fractions or total concentrations.

Recommendations: The only way to determine the reducible fraction is to find out what a certain amount really looks like. Possible, one way to do this, would be to make up a solution of distilled water and a simple salt such as NaCl or KCl. There can be no trace metal or organic

contamination from the reagents and there can be nothing that will irreversibly complex the trace metals. Reproducibility should be checked for varying temperature, salinity and pH. Having established these conditions, scans should be run for all situations (i.e. varying deposition time, sensitivity, etc.) and basically then, all the sample peaks could be compared to the scan having matching conditions.

Another possible way would be to run a scan on the sample and then by acid digestion or UV irridation (or whatever is necessary) break up the trace metal compounds and oxidize the organics. Then when the spike is added, the metal ions will stay reducible because there is nothing to complex with.

Primary Reference: P.L. Brezonik, P.A. Brauner, W. Strumm.

Trace Metal Analysis by Anodic Stripping

Voltammetry: Effect of Sorption by Natural and Model  
Organic Compounds. Water Res. 10: 605-612.

PETIT MANAN POINT 1979

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>CU ppb</u>
7-27-79	0.00	0.788	0.518
9-5	0.00	0.599	0.00
9-8	0.00	1.01	0.344
9-11	0.00	1.86	0.735
9-23	0.00	0.00	0.00
9-14	0.123	0.631	0.00
9-17	0.00	0.00	1.17
8-5	1.37	2.27	0.431
9-26	0.172	1.19	0.30
7-13	0.270	1.32	0.00
9-30	0.126	0.373	0.00
7-21	0.07	1.40	0.778
7-9	0.169	16.88	0.00
7-23	0.04	0.560	0.00
7-25	0.114	1.54	0.767
8-8	0.311	1.12	

\*Samples were stored frozen

PETIT MANAN POINT (PMP) 1978

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
7-13-1978	0.00	0.683	1.12
7-14	0.00	0.583	0.00
7-16	0.00	1.94	0.528
7-17	0.487	1.73	5.15
7-19	0.00	0.769	0.530
7-21	0.00	0.412	0.788
7-23	0.083	0.293	0.00
7-24	0.00	1.29	0.00
7-26	0.00	0.458	0.00
7-28	0.00	0.490	0.911
7-30	0.250	1.13	0.00
8-1	0.00	0.275	0.00
8-3	0.00	0.412	0.00
8-5	0.084	0.458	0.00
8-4	0.042	0.234	0.381
8-9	0.021	0.469	0.00
8-11	0.042	0.557	0.00
8-13	0.104	0.352	0.00
8-15	0.00	0.586	0.381
8-17	0.063	0.205	0.190
8-19	0.104	0.446	0.457
8-21	0.250	1.40	0.480
8-23	0.133	1.23	2.90

PETIT MANAN POINT (PMP) 1979 - continued

<u>Date</u>	<u>CD ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
8-26-79	0.00	1.09	0.00
8-28	0.079	0.939	0.302
8-30	0.00	0.797	0.490
9-3	0.026	0.531	0.321
9-5	0.053	0.531	1.10
9-7	0.0	0.836	0.00
9-11	0.053	0.817	0.059
9-17	0.00	0.408	0.00
9-19	0.00	0.327	0.382
9-20	0.221	0.648	0.321
9-23	0.131	0.449	0.326
9-27	0.053	0.643	

\*All samples were acidified

PROVINCETOWN, MASS. (P) 1978 - 1979

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
3-15-78	0.330	0.940	1.30
3-22	0.190	0.970	1.80
3-29	0.180	0.450	1.50
4-5	0.00	2.16	1.84
4-12	0.379	0.304	0.853
4-17	0.300	3.07	1.37
4-26	0.808	4.13	4.21
5-3	0.00	2.83	0.909
5-10	0.563	1.46	0.00
5-17	0.00	-	1.81
5-24	0.442	3.75	2.80
5-31	0.309	8.26	2.36
6-7	0.00	2.45	0.814
6-14	0.137	1.84	1.62
6-21	0.942	8.41	0.967
6-28	0.518	1.78	0.00
7-5	0.452	0.749	1.12
7-12	0.250	5.50	0.946
7-19	0.230	1.40	2.50
8-3	3.29	1.86	0.533
8-9	0.480	6.60	3.53
8-16	0.120	3.30	0.372
8-23	0.650	3.60	2.57

PROvincetown, MASS. (P) 1978-1979 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
8-31-1978	0.700	4.17	2.99
9-13	0.998	2.28	4.20
9-20	1.54	3.00	6.60
9-28	0.910	1.80	0.923
10-5	0.390	2.69	1.01
10-11	0.400	7.59	2.20
10-18	0.980	10.87	5.79
10-26	0.392	7.11	8.29
11-29	0.336	7.56	0.727
12-8	0.289	1.05	0.00
12-13	0.168	1.34	0.00
12-21	0.213	0.825	1.28
12-27	0.536	1.85	0.760
1-3-1979	0.00	2.83	0.326
1-10	0.285	13.33	0.00
1-17	0.321	1.73	0.00
1-24	0.357	2.05	0.00
1-31	0.178	2.46	0.00
2-8	0.393	0.725	0.00
2-2	0.035	0.797	0.00
3-7	2.14	2.19	0.00
3-22	0.00	2.56	1.15

PROvincetown, MASS. (P) 1978 - 1979 - Continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
3-28-1979	0.194	3.19	1.10
4-4	0.194	0.799	0.00
4-25	0.535	1.86	2.83
5-3	0.392	2.03	2.00
5-10	0.214	1.48	3.83
5-16	0.225	1.51	1.87

\*All samples were frozen

## MCKOWN POINT (MP) 1978

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
7-3-1978	0.028	0.180	0.928
7-6	0.056	1.06	1.75
7-11	0.028	1.80	0.948
7-13	0.079	0.301	1.05
7-18	0.040	0.356	0.405
7-20	0.099	1.04	1.64
7-25	0.119	0.575	0.619
7-27	0.020	0.575	0.619
8-1	0.079	0.438	0.953
8-3	0.020	1.37	2.33
8-8	0.040	0.657	0.595
8-10	0.099	0.309	1.73
8-15	0.028	0.574	7.03
8-17	0.083	1.77	0.00
8-22	0.083	0.530	0.00
8-24	0.111	0.441	0.00
8-29	0.139	0.618	0.973
8-31	0.056	1.10	0.379
9-6	0.222	0.397	0.00
9-8	0.00	0.441	0.00
9-13	0.00	1.46	0.00
9-15	0.083	0.971	0.811
9-18	0.139		

\* Samples were stored acidified

MCKOWN POINT (MP) 1978

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
5-25-1978	-	-	0.289
5-30	-	-	0.00
6-1	1.45	2.51	1.14
6-6	2.88	1.27	0.610
6-8	-	-	1.06
6-13	0.643	1.54	6.10
6-15	0.964	2.29	0.970
6-20	1.13	2.32	1.09
6-27	1.75	0.449	0.638
6-29	-	-	0.435
9-21	0.125	0.877	0.580
9-22	-	-	0.435
9-25	0.00	0.065	0.203
9-28	0.00	1.01	0.493

\*Samples were stored frozen

## NEWAGEN (NW) 1979

<u>Date</u>	<u>Cd ppb</u>	<u>PB ppb</u>	<u>Cu ppb</u>
1-4-79	0.105	0.690	0.893
1-11	0.00	-	0.00
1-19	0.156	0.295	0.00
2-1	1.69	0.737	0.00
2-9	0.245	0.295	0.864
2-15	2.59	0.652	0.853
2-22	0.531	3.12	2.68
3-1	0.406	0.765	1.28
3-8	0.313	0.765	0.548
3-23	0.00	0.00	0.592
4-6	0.00	-	0.563
4-20	0.062	0.495	0.813
4-26	0.00	-	0.438
5-14	0.302	0.748	0.512
5-15	0.00	1.27	0.00
5-17	0.00	0.00	0.00
5-18	0.00	0.00	0.308
5-21	0.174	0.953	0.00
5-22	0.00	0.00	0.00
5-23	0.00	2.35	0.551
5-24	0.290	0.715	0.00
5-25	0.00	0.00	0.00
5-26	0.06	0.446	0.710

NEWAGEN (NW) 1979 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>PB ppb</u>	<u>Cu ppb</u>
5-28-79	0.00	0.465	0.443
5-29	0.00	0.883	0.00
5-30	0.00	0.00	0.441
6-1	0.164	1.75	0.521
6-2	0.211	1.81	0.413
6-3	0.07	0.446	0.418
6-4	0.080	0.643	0.00
6-5	0.00	1.10	0.640
6-6	0.263	0.643	0.00
6-7	0.00	1.06	0.00
6-8	0.195	1.22	1.27
6-10	0.269	0.815	1.08
6-11	0.167	1.49	1.20
6-12	0.202	1.30	0.692
6-13	0.445	3.55	4.67
6-15	0.269	1.26	0.845
6-16	0.312	1.94	0.00
6-17	0.078	1.04	0.00
6-18	0.221	1.81	0.501
6-19	0.185	0.321	0.00
6-20	0.00	.963	0.00
6-21	0.185	0.723	0.580

NEWAGEN (NW) 1979 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
6-22-79	0.850	4.40	1.03
6-24	0.250	1.37	1.91
6-26	0.161	1.44	0.250
6-27	0.221	1.65	1.67
6-29	0.221	0.604	0.543
6-30	0.680	1.90	1.02
7-1	0.098	9.03	0.00
7-2	0.115	1.86	0.417
7-3	0.098	1.13	0.190
7-4	0.110	0.815	0.00
7-5	0.017	0.184	0.00
7-13	0.159	0.884	1.91
7-14	0.195	1.49	1.12
7-16	0.221	1.63	0.919
7-17	0.148	1.60	0.00
7-21	0.054	0.281	0.00
7-22	5.55	1.17	0.00
7-23	0.524	0.643	0.00
7-24	0.213	1.01	0.312
7-25	0.524	0.00	0.00
7-26	0.181	0.997	0.418
7-27	0.837	1.48	2.66
7-28	0.090	0.155	0.00

NEWAGEN (NW) 1979 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>PB ppb</u>	<u>Cu ppb</u>
7-30-79	0.259	0.814	1.34
7-31	0.233	0.845	0.542
8-1	0.250	0.756	1.76
8-2	0.00	0.408	0.00
8-6	2.06	0.971	1.55
8-7	0.202	1.89	0.922
8-8	0.151	0.680	0.00
8-10	0.00	0.00	0.00
8-12	0.111	1.66	0.824
8-13	0.347	1.41	0.496
8-14	0.00	0.408	0.00
8-15	0.00	1.18	0.551
8-16	0.00	1.47	0.882
8-20	0.00	0.380	0.00
8-24	0.161	0.526	0.125
8-29	0.237	0.803	0.997

\*samples were stored frozen

## NEWAGEN (NW) 1978

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
4-12-78	0.146	0.371	1.56
4-21	0.188	2.95	2.44
4-28	0.156	2.67	1.90
5-18	0.312	1.48	1.08
5-25	0.187	0.758	0.563
6-2	-	-	0.00
6-5	0.341	0.863	1.12
6-6	0.971	2.32	0.983
6-7	0.299	0.842	0.269
6-8	0.185	2.89	1.28
6-9	0.124	0.290	0.537
6-12	0.249	0.232	0.314
6-13	0.473	1.25	0.314
6-14	0.00	5.13	7.69
6-15	0.523	0.377	0.00
6-16	0.069	0.769	0.964
6-19	0.998	2.15	0.313
6-21	0.270	1.51	0.563
6-22	0.149	0.261	0.671
6-23	0.00	0.272	3.13

\* samples were stored frozen

NEWAGEN (NW) 1978 - continued

6-26-78	0.208	-	1.00
6-27	0.156	1.58	3.13
6-28	0.443	0.365	0.00
6-29	0.469	1.14	1.88
9-21	0.00	0.00	0.432
9-22	0.156	0.324	0.864
9-25	0.00	0.088	0.00
9-26	0.00	0.501	0.00
9-27	0.037	0.115	0.00
9-28	0.00	-	0.540
9-29	-		0.446
10-5	0.156	0.383	0.00
10-12	0.067	0.413	1.08
10-19	0.281	0.709	0.609
11-17	0.055	0.230	0.00
11-22	0.129	0.402	0.529
11-30	0.314	2.19	4.01
12-7	0.222	1.96	0.982
12-14	0.00	0.097	0.00
12-21	-	0.369	0.00
12-28	0.185	0.245	0.502

\* samples were stored frozen

## NEWAGEN (NW) 1978

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
6-30-78	0.00	0.482	1.40
7-3	0.00	0.593	0.00
7-5	0.099	0.223	0.425
7-6	0.133	0.408	0.00
7-7	0.00	0.354	0.00
7-10	0.079	0.248	0.00
7-11	0.027	0.248	1.60
7-12	0.106	0.00	0.456
7-13	0.079	0.036	0.608
7-14	0.158	0.354	0.380
7-17	0.079	0.177	0.951
7-18	0.083	0.317	0.00
7-19	0.028	0.202	0.574
7-20	0.00	0.260	0.00
7-21	0.028	0.433	0.00
7-24	0.00	0.189	0.00
7-25	0.106	0.472	0.00
7-26	0.035	0.472	0.00
7-27	0.035	0.142	0.00
7-28	0.142	0.425	1.49
7-31	0.142	0.236	0.416
8-1	0.079	0.374	0.719

## NEWAGEN (NW) 1978 - continued

<u>Date</u>	<u>Cd pbb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
8-2-78	0.063	0.214	0.569
8-3	0.032	0.294	0.180
8-4	0.063	0.828	0.00
8-7	0.032	0.267	0.00
8-8	0.079	0.267	0.00
8-9	0.079	0.427	0.360
8-10	0.00	0.240	1.05
8-11	0.00	0.150	0.323
8-14	0.00	0.275	0.00
8-15	0.093	0.175	1.37
8-16	0.037	0.250	0.223
8-17	0.00	0.200	0.00
8-18	0.056	1.03	0.049
8-19	0.148	0.702	2.81
8-22	0.092	0.376	1.48
8-23	0.184	1.28	1.43
8-24	0.092	0.482	0.00
8-25	0.062	0.535	0.00
8-28	0.092	0.641	0.00
8-29	0.123	0.535	0.00
8-31	0.107	0.639	1.13
9-1	0.053	0.545	0.614
9-5	0.018	0.331	0.593

## NEWAGEN (NW) 1978 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
9-6-78	0.071	0.284	0.00
9-7	0.036	0.426	0.00
9-8	0.142	0.521	0.00
9-11	0.339	1.16	0.553
9-13	0.053	0.260	0.00
9-14	0.125	0.710	0.00
9-15	0.079	0.339	2.04
9-18	0.019	0.498	1.50
9-19	0.039	0.279	0.074
9-20	0.079	0.737	0.056

\* samples were acidified

## MONHEGAN 1977

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
6-16-77	0.19	0.32	3.23
6-17	0.22	2.25	3.45
6-18	0.24	0.17	1.04
6-20	0.11	0.13	0.88
6-22	0.09	0.72	3.15
6-23	0.00	0.25	1.60
6-24	0.08	0.24	1.40
6-26	0.00	0.12	2.10
6-27	0.00	0.27	1.90
6-28	0.27	0.32	1.88
6-29	0.22	0.35	1.70
6-30	0.36	0.26	1.90
7-1	0.12	0.23	2.40
7-2	0.24	0.17	0.97
7-3	0.83	0.65	1.15
7-4	0.85	0.56	0.98
7-5	0.36	0.00	2.22
7-6	0.08	0.00	1.62
7-7	0.28	0.33	1.50
7-8	0.38	0.26	1.60
7-9	0.48	1.37	1.20
7-10	0.38	2.94	3.47
7-11	0.22	2.94	3.47

## MONHEGAN 1977 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
7-12-77	0.74	0.63	2.64
7-13	0.86	0.31	4.15
7-15	0.14	0.13	1.40
7-16	0.15	0.27	1.43
7-17	0.09	0.24	2.60
7-18	0.47	1.30	2.60
7-19	0.00	0.00	1.30
7-20	0.07	0.51	0.50
7-21	0.22	0.28	1.42
7-22	0.00	0.00	1.21
7-26	0.21	0.23	0.64
7-27	0.07	0.51	0.50
7-28	0.00	0.26	0.60
7-29	0.14	0.10	1.28
7-30	0.08	0.20	2.70
7-31	0.00	0.24	1.47
8-1	0.09	0.18	0.91
8-2	0.00	0.80	0.95
8-3	0.01	0.13	1.06
8-4	0.00	0.15	2.07
8-5	0.81	0.21	-
8-7	0.10	0.32	1.57
8-8	0.64	0.76	-

## MONHEGAN 1977 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
8-9-77	0.45	0.27	1.45
8-10	0.00	0.39	1.58
8-11	0.58	0.51	1.28
8-14	0.42	0.34	1.35
8-18	0.06	0.22	0.78
8-21	0.00	0.17	4.23
8-22	0.42	1.36	3.80
8-23	0.00	0.00	1.13
8-24	0.00	0.12	1.30
8-25	0.00	0.28	1.90
8-26	0.08	0.07	1.37
8-27	0.08	0.38	1.74
8-28	0.14	1.01	1.43
8-29	0.12	0.01	0.83
8-30	0.19	0.28	0.70
8-31	0.00	0.22	1.60
9-2	0.00	0.41	3.70
9-4	0.18	0.28	1.66
9-5	0.12	0.35	1.93
9-6	0.00	0.70	5.90
9-8	0.01	0.23	0.00
9-9	0.03	0.16	1.37
9-10	0.07	0.17	1.60

MONHEGAN 1977 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
9-11	0.08	0.37	0.83
9-15	0.03	0.28	1.00
9-16	0.13	0.14	0.80

\*samples stored frozen

## MONHEGAN 1978

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
6-7-78	0.56	0.99	1.19
7-9	0.00	4.40	1.08
7-10	-	-	0.00
7-11	-	-	0.00
7-12	0.092	6.53	9.15
7-13	0.00	0.216	1.18
7-14	0.00	2.16	2.57
7-15	0.354	3.78	4.56
7-16	0.062	0.698	0.579
7-17	0.00	2.10	0.616
7-18	0.333	3.04	2.81
7-19	0.354	-	1.25
7-20	0.00	0.00	1.48
7-21	0.284	1.36	2.36
7-22	0.35	2.16	3.82
7-23	0.208	1.74	1.94
7-24	0.00	0.413	0.415
7-25	0.289	1.26	0.609
7-26	0.00	12.96	2.74
7-27	0.186	0.782	0.00
7-28	0.084	0.00	0.783
7-29	0.145	0.271	0.00
7-30	0.236	12.37	1.33

## MONHEGAN 1978 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
7-31	0.00	0.094	0.00
8-1	0.00	0.095	0.00
8-2	0.37	11.41	2.13
8-3	0.118	1.57	0.354
8-4	0.167	1.47	0.00
8-5	0.057	0.569	0.00
8-6	0.00	0.262	0.00
8-7	0.157	0.523	3.33
8-8	0.054	8.29	1.39
8-9	5.30	8.52	7.41
8-10	0.00	0.248	0.909
8-11	0.112	0.428	1.18
8-12	0.114	4.16	1.32
8-13	0.207	1.29	1.06
8-14	0.112	0.500	0.00
8-15	0.108	1.43	3.26
8-16	0.00	0.00	0.00
8-17	0.00	0.188	0.00
8-18	0.567	1.75	3.83
8-20	0.00	-	0.00
8-21	0.057	0.133	0.00
8-23	0.00	7.42	0.00
8-24	0.029	1.19	0.00

## MONHEGAN 1978 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
8-25	0.00	0.00	0.00
8-26	0.00	0.00	1.06
8-27	0.00	0.723	2.34
8-28	0.00	5.41	1.97
8-29	0.204	1.82	0.00
8-30	0.00	0.928	0.00
8-31	0.00	-	0.579
9-2	0.292	0.928	1.65
9-3	0.00	0.00	0.00
9-4	-	-	0.773
9-5	0.00	0.467	0.870
9-6	0.091	17.70	2.31
9-7	0.115	0.339	1.06
9-8	0.208	0.467	1.06
9-10	0.066	0.399	0.319
9-11	0.00	3.67	3.49
9-12	0.00	0.469	0.354
9-13	0.098	1.12	4.18
9-14	0.00	0.262	0.00
9-15	0.00	0.928	0.00
9-16	0.233	1.04	0.056
9-17	0.00	9.91	2.98
9-18	0.029	0.334	0.00

MONHEGAN 1978 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
9-19	0.305	4.90	1.79
9-20	0.104	0.785	0.731
9-21	0.00	5.46	1.35
9-22	0.00	0.631	0.00
9-23	0.00	0.334	0.112
9-24	0.058	0.631	0.00
9-25	0.058	0.483	0.503
9-26	0.00	0.334	0.00
9-27	0.07	0.942	0.262
9-28	0.00	0.557	0.00

\* samples were stored frozen

## MONHEGAN 1978 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
6-7-78	0.18	2.01	1.24
6-8	0.170	0.659	0.00
6-9	0.056	0.375	0.00
6-10	0.094	0.468	0.166
6-11	0.227	0.312	0.166
6-12	0.113	0.219	0.00
6-13	0.209	0.570	0.00
6-14	0.139	0.482	0.00
6-15	0.278	0.833	0.628
6-16	0.348	0.833	1.35
6-17	0.17	0.37	1.26
6-18	0.54	2.39	0.84
6-19	0.03	0.98	2.26
6-20	0.07	1.20	2.69
6-21	0.15	0.67	0.64
6-22	0.10	0.26	0.55
6-23	0.12	0.83	4.16
6-24	0.09	0.09	3.14
6-25	0.16	0.37	8.86
6-26-78	0.10	0.31	2.47
6-27	0.05	0.40	1.34
6-28	0.11	0.31	1.65
6-29	0.21	0.45	2.35

## MONHEGAN 1978 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
7-1	0.035	0.219	0.00
7-2	0.104	0.702	1.44
7-3	0.313	0.307	0.00
7-4	0.278	4.34	3.59
7-5	0.00	0.439	0.00
7-6	0.00	0.00	0.00
7-7	0.035	0.351	0.00
7-8	0.243	0.351	0.00
7-9	0.00	0.594	0.798
7-10	0.209	0.219	0.00
7-11	0.062	0.378	0.00
7-12	0.00	2.59	2.52
7-13	0.062	0.258	0.00
7-14	0.00	2.18	2.90
7-15	0.094	0.420	0.00
7-16	0.219	0.798	0.00
7-17	0.125	0.462	0.00
7-18	0.031	0.168	0.00
7-19	0.00	0.714	0.00
7-20	0.062	0.294	0.00
7-21	0.156	0.294	0.00
7-22	0.00	1.34	1.70

## MONHEGAN 1978 - continued

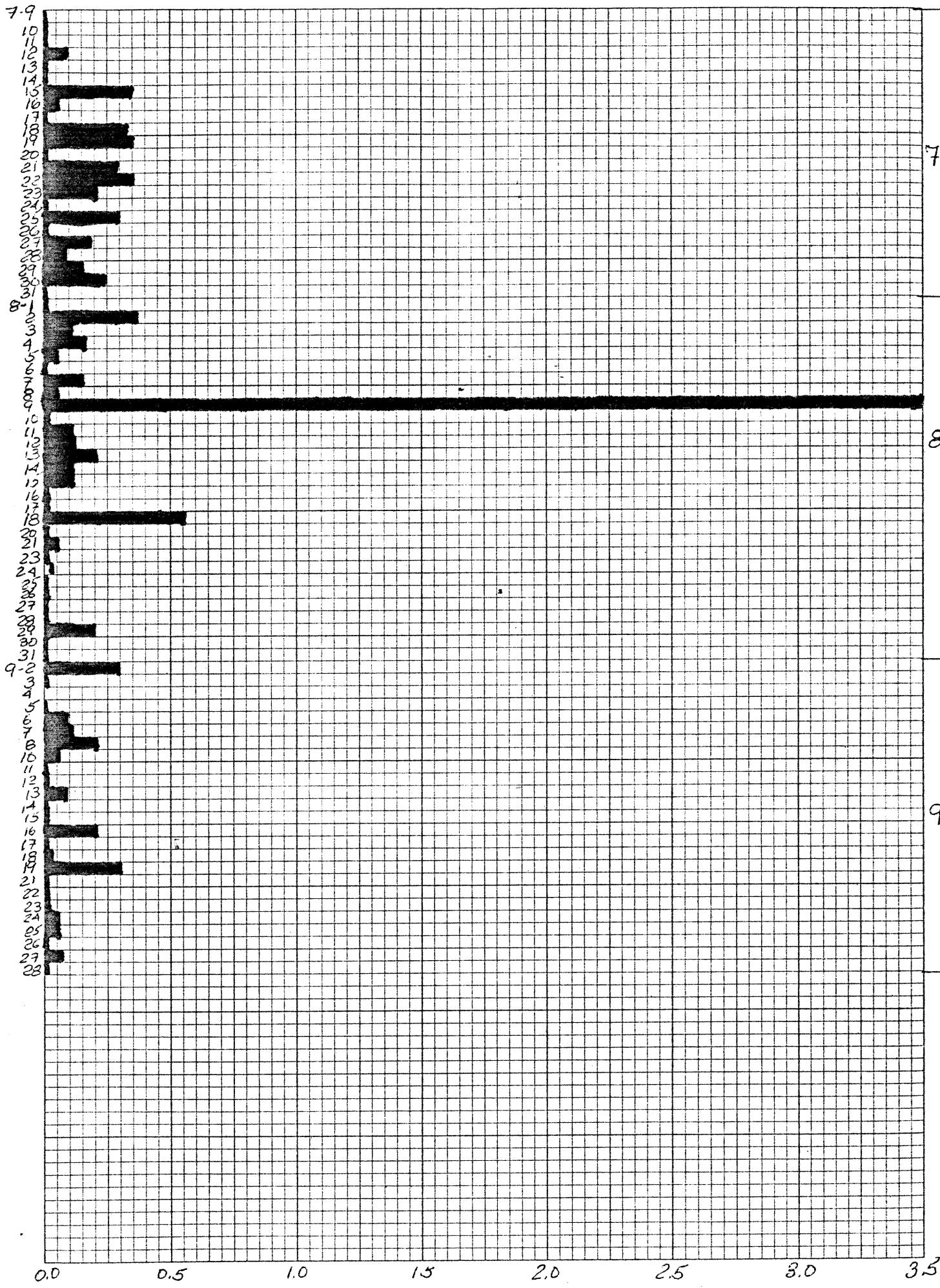
<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
7-23-78	0.095	0.389	0.00
7-24	0.00	0.871	4.95
7-25	0.719	0.420	0.00
7-26	0.000	0.380	1.20
7-27	0.187	0.210	0.00
7-28	0.164	0.119	0.00
7-29	0.218	0.697	1.04
7-30	0.00	0.551	1.10
7-31	0.146	0.648	0.031
8-1	0.146	0.072	0.086
8-3	0.128	0.151	0.00
8-4	0.110	0.001	0.00
8-5	0.055	0.117	0.00
8-6	0.073	0.606	0.00
8-7	0.073	0.049	0.00
8-8	0.00	0.459	0.404
8-9	0.30	0.993	1.40
8-10	0.128	0.072	0.024
8-11	0.146	0.114	0.00
8-12	0.421	0.151	0.024
8-13	0.677	0.140	0.078
8-14	0.091	0.095	0.00
8-15	0.151	0.882	0.263

## MONHEGAN 1978 - continued

<u>Date</u>	<u>Cd ppb</u>	<u>Pb ppb</u>	<u>Cu ppb</u>
8-16-78	0.133	1.15	0.00
8-17	0.114	0.389	0.00
8-20	0.056	0.593	0.00
8-21	0.114	1.38	0.00
8-22	0.122	0.652	0.00
8-23	0.151	0.652	0.00
8-24	0.114	0.586	0.00
8-25	0.189	1.24	0.00
8-26	0.151	0.849	0.00
9-2	0.208	1.84	0.919
9-3	0.151	0.718	0.00
9-4	0.133	0.882	0.00
9-5	0.151	0.816	0.00
9-6	0.167	0.960	0.589
9-8	0.130	1.07	0.00
9-9	0.074	0.621	0.00
9-10	0.056	0.480	0.00
9-11	0.130	0.536	0.00
9-12	0.149	0.621	0.00
9-13	0.111	0.762	1.28
9-14	0.111	0.621	0.00
9-15	0.130	0.932	0.00

\* samples were acidified

# Monhegan 1978



Monhegan 1977

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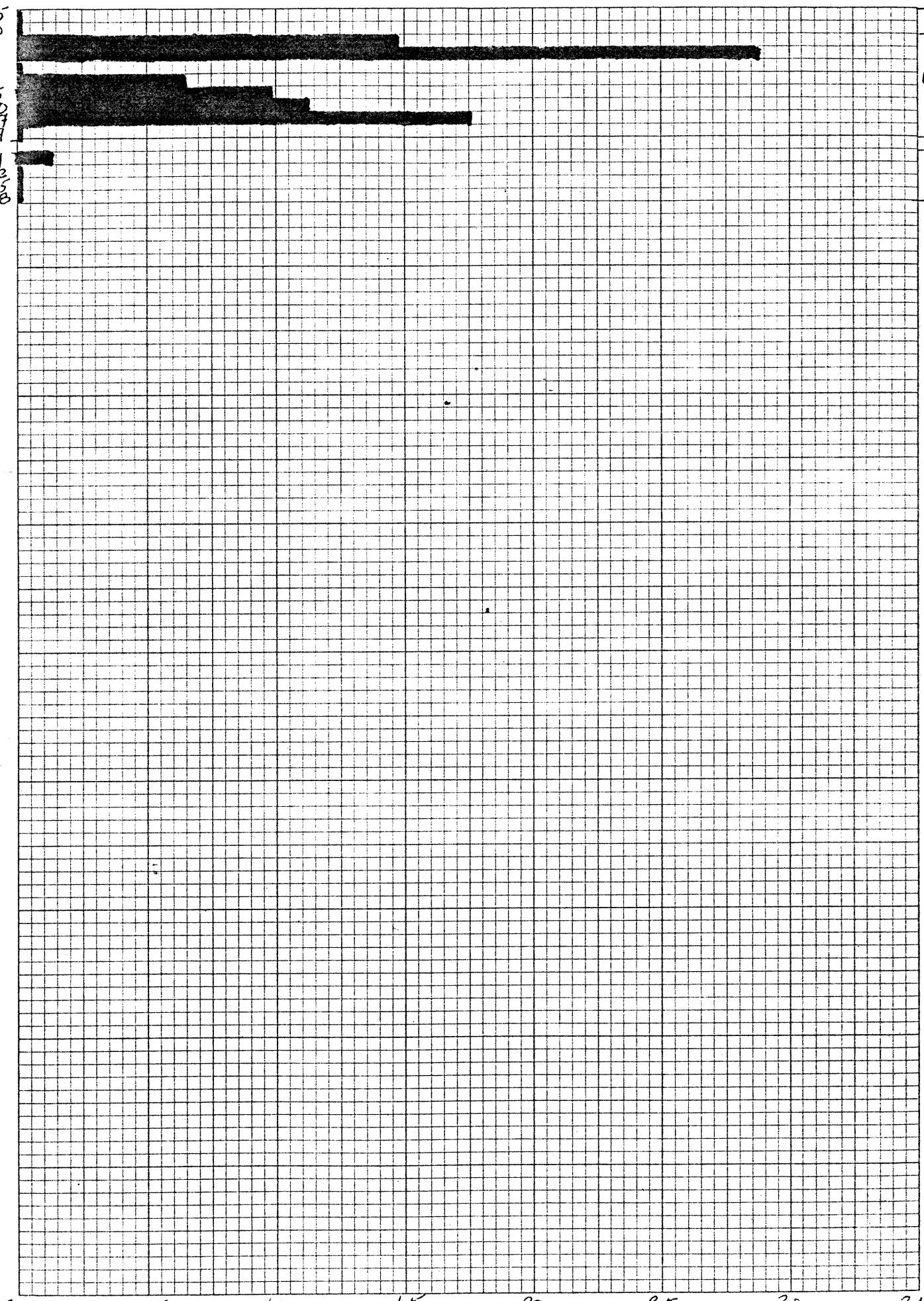
Mckown Pt. 1978

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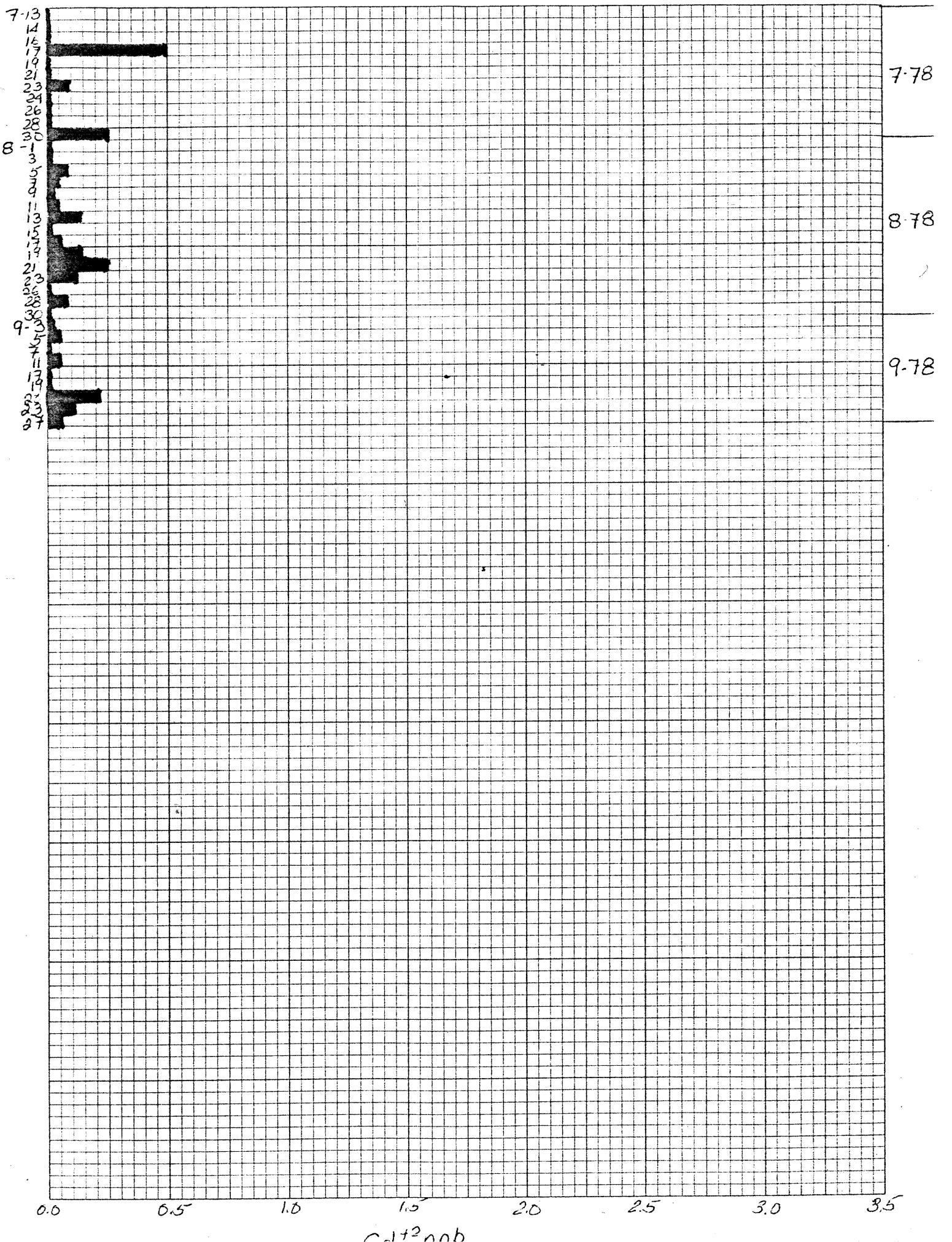
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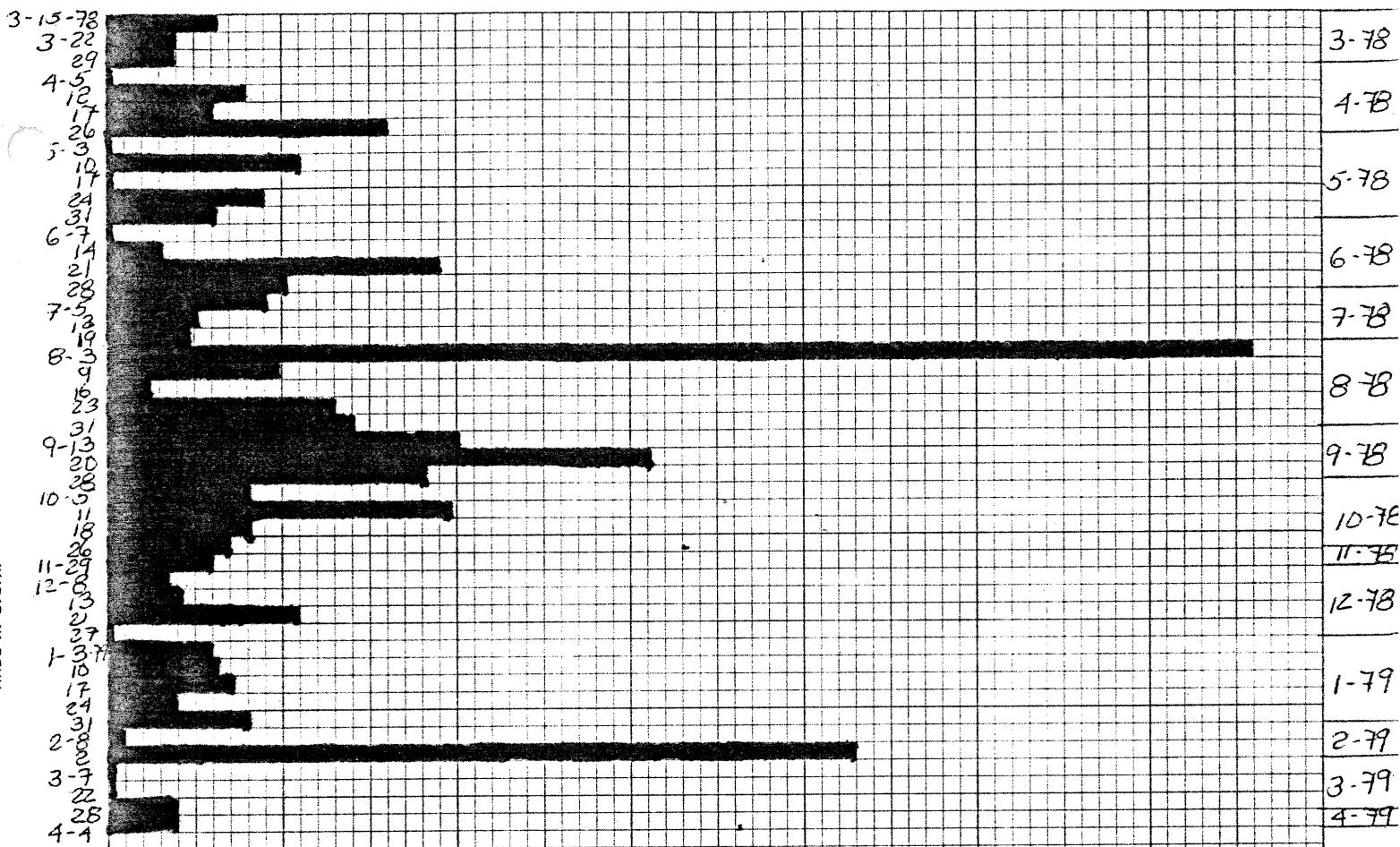
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Petit Manan Pt 1978



Provincetown 1978 + 1977



Monhegan 1978

7-78

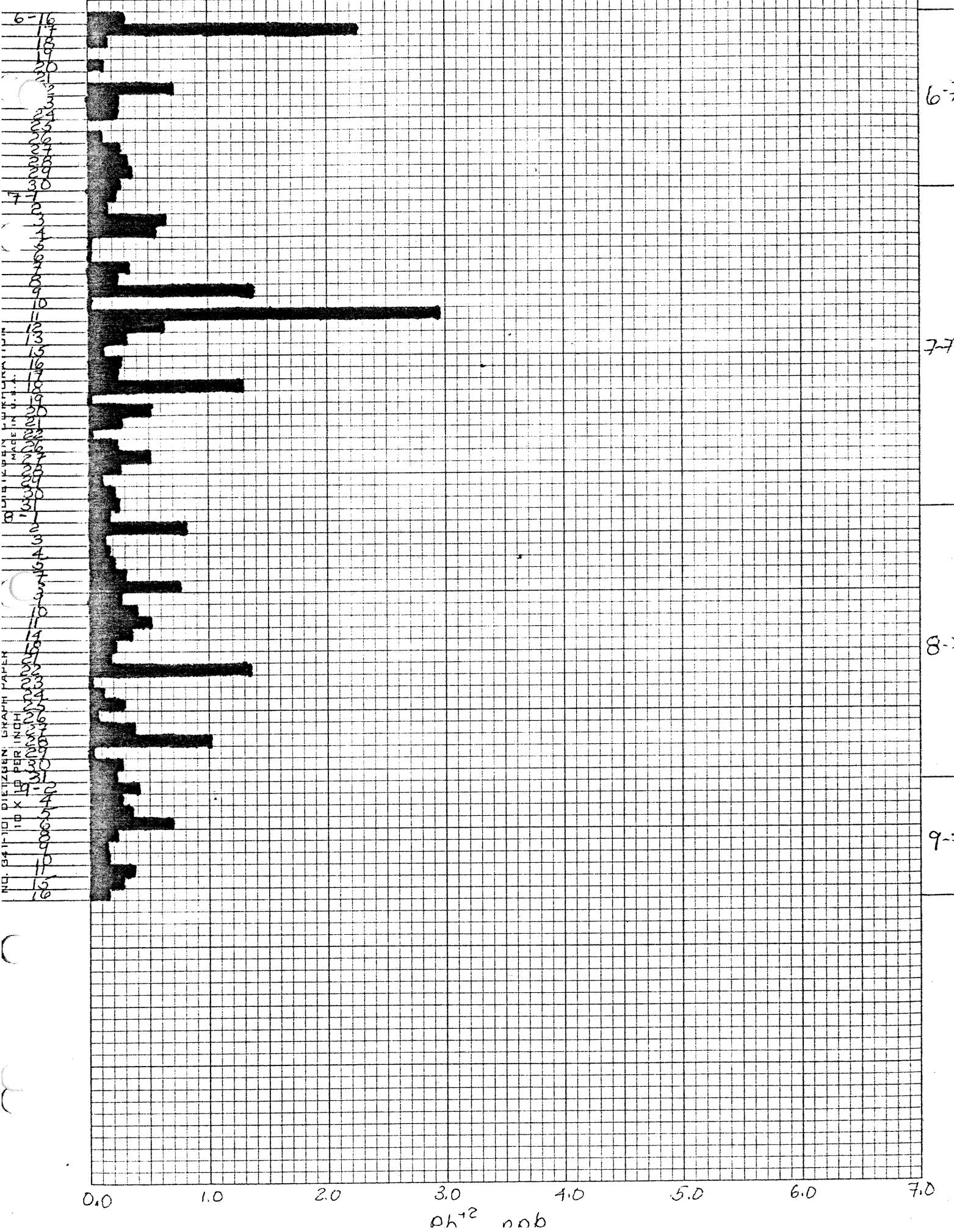
878

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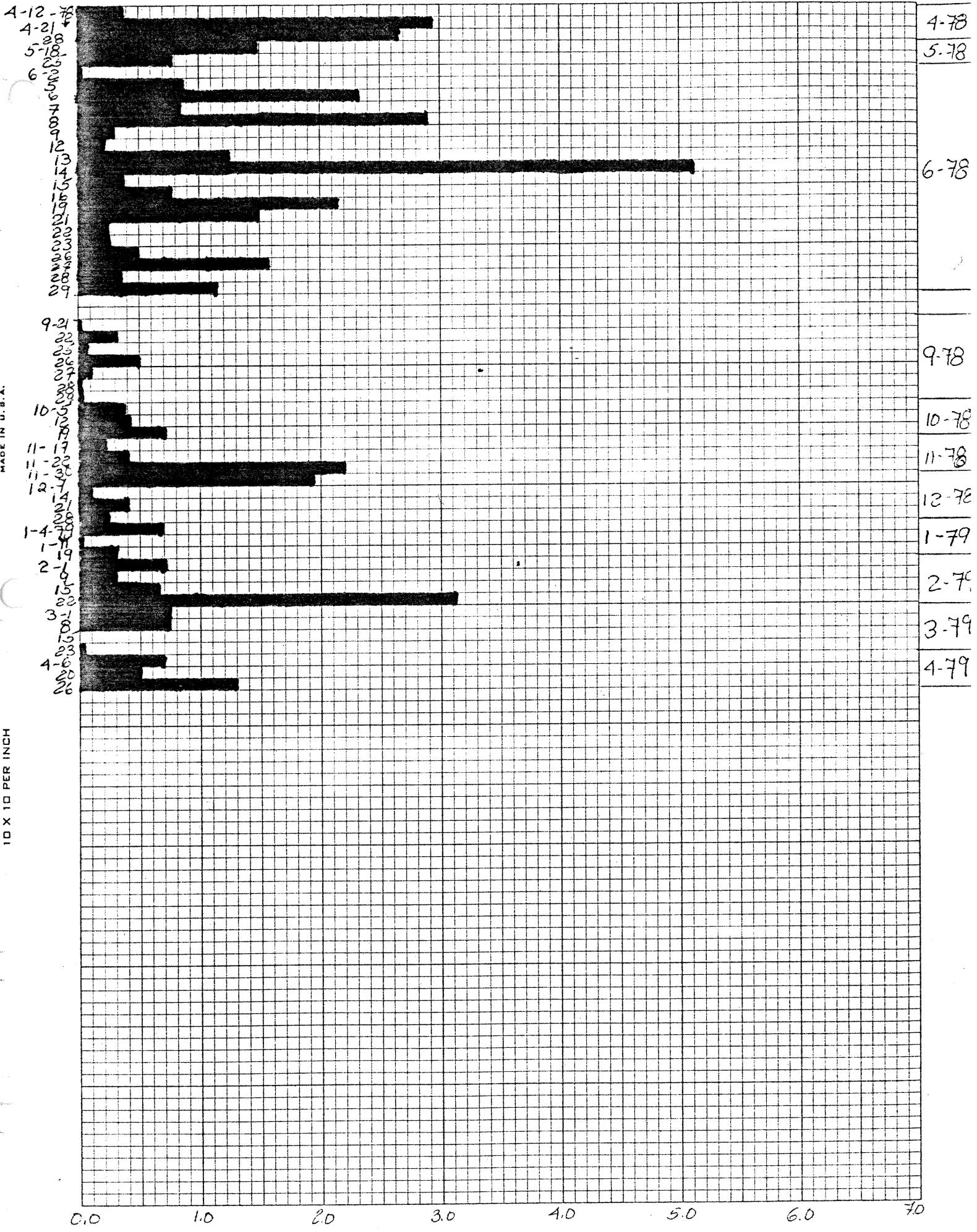
MADE IN U. S. A.

10 X 10 PER INCH

Monhegan 14 ft



# Newagen 1978 - 1979



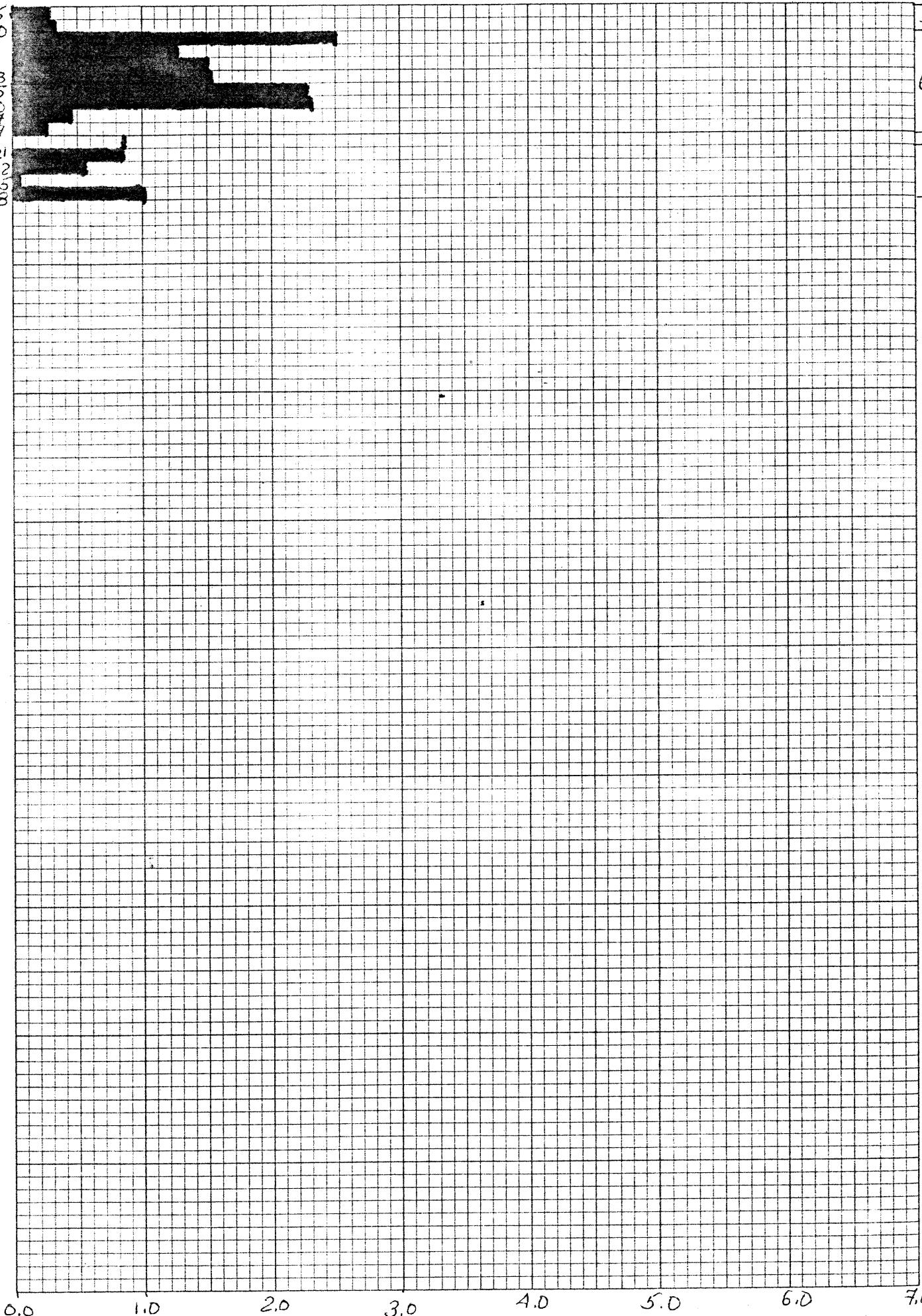
McKown Pt. 1978

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WIELEGGEN LUMPFWURZEN  
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NU. 341-11 DIE-ZAGEN GRAPH PAPER  
10 X 10 PER INCH

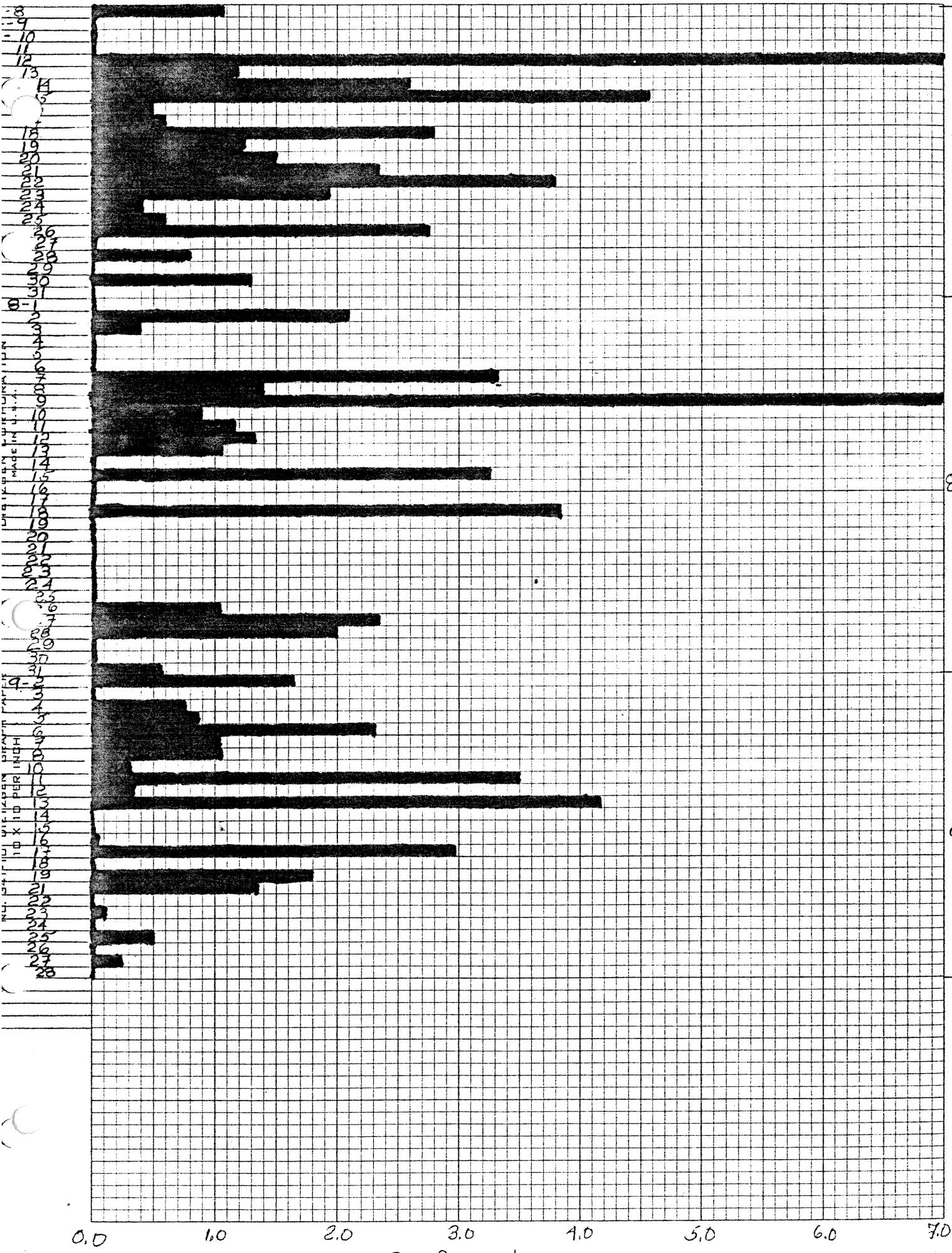
## Provincetown 19+8+17+7

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Monnegan 1770



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Cu,72 ppb

# Monhegan 1977

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Mckown Pt. 1978

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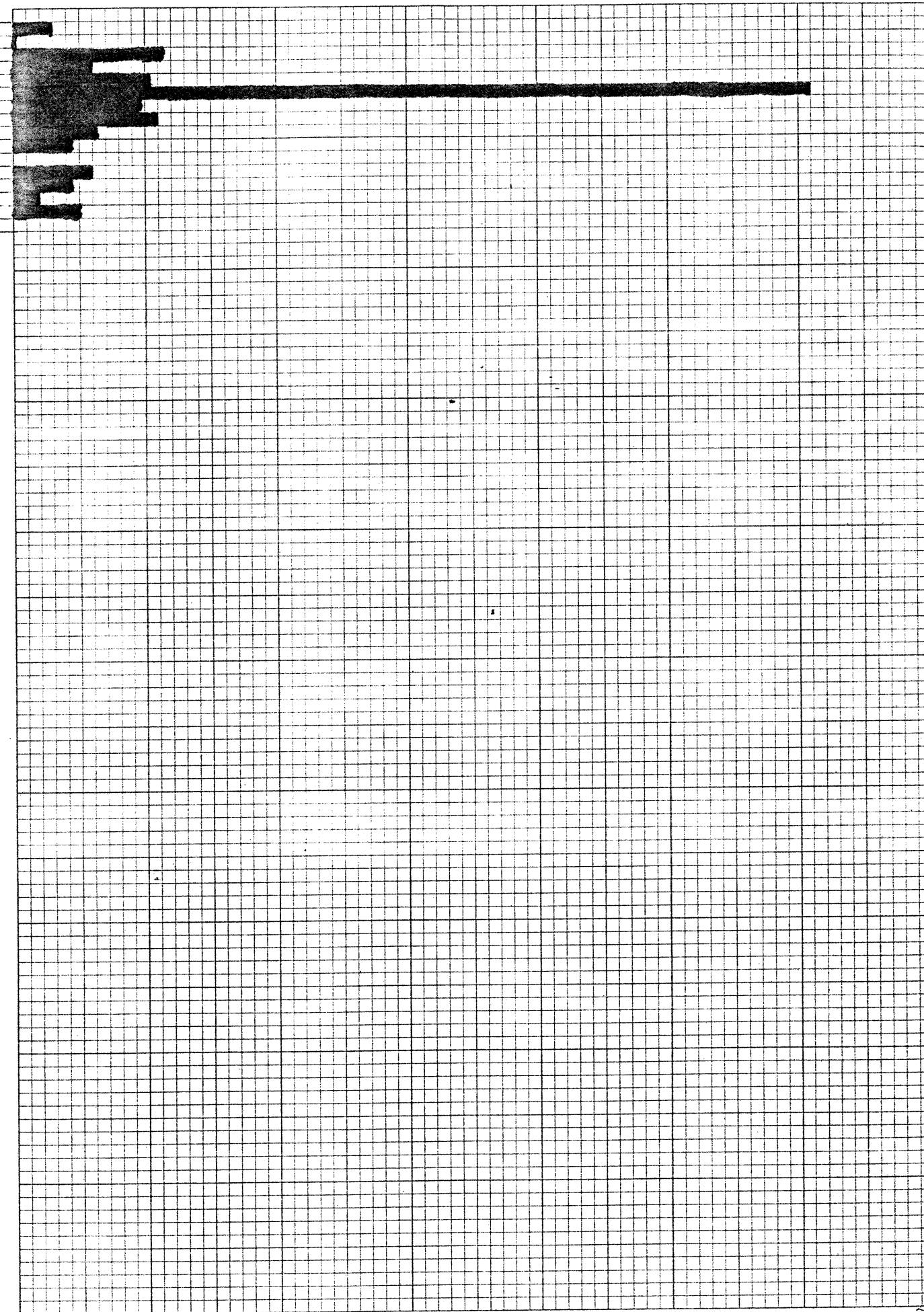
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MADE IN U.S.A.

NW. 34110 WIEZEN KRAFT PAPER  
10 X 10 PER INCH



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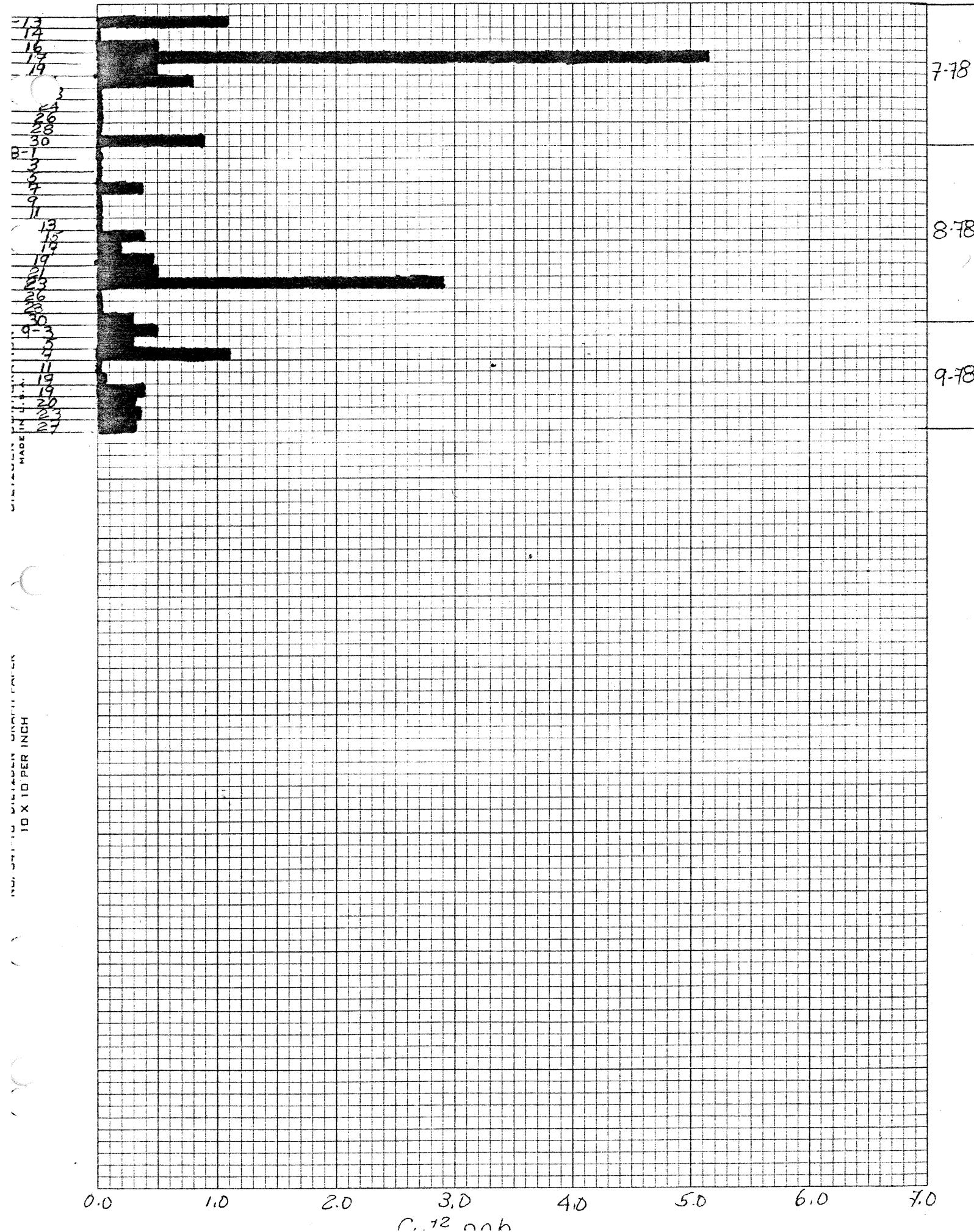
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Petit Manan Pt. 1770



## PROVINCETOWN 14 TO + 1777

